

Study on the Influence of Different Ratios of CO₂ and N₂ on Extra-Heavy Oil Reservoirs Developed by Multi-Thermal-Fluid Huff-Puff[#]

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ABSTRACT

Flue gas assisted cyclic steam stimulation (CSS) has gradually gained prominence in the development of extra-heavy oil reservoirs. The flue gas produced by fuel combustion is primarily composed of CO₂, N₂, and steam, and it is injected directly into extra-heavy oil reservoirs to enhance oil recovery. This method is also known as multi-thermal-fluid (MTF) huff-puff technology, which is an important CCUS technology. In this paper, the effects of different CO₂ and N₂ ratios on the development of extra-heavy oil reservoirs are analyzed and discussed. To begin, the viscosity and components of crude oil are obtained through physical experiment. Second, the effects of different CO₂ and N₂ ratios on crude oil viscosity and phase state are analyzed using the phase equilibrium model and viscosity model. Finally, numerical simulation is used to optimize the CO₂ and N₂ ratios. The results show that the higher the CO₂ ratio, the lower the viscosity of the crude oil after MTF huff-puff. The phase diagram gradually narrows as CO₂ ratios rise while N₂ ratios fall. Because the primary function of CO₂ is to reduce viscosity and the primary function of N₂ is to improve formation energy, the optimal ratio of CO₂ and N₂ is 7:3.

Keywords: Extra-heavy oil reservoir, Multi-thermal-fluid huff-puff, Optimal ratios of CO₂ and N₂, PVT simulation, Numerical simulation.

NONMENCLATURE

Abbreviations

MTF	Multi-thermal-fluid
CSS	Cyclic steam stimulation
SF	Steam flooding
SAGD	Steam assisted gravity drainage
ISC	In-situ combustion

Symbols

a_m	Empirical coefficient
b_m	Empirical coefficient
K_i	Phase equilibrium constant

k_{ij}	Binary interaction coefficient
p	Pressure
T	Temperature
p_c	Critical pressure
T_c	Critical temperature
x_i	Liquid phase fraction
y_i	Gas phase fraction
Z	Deviation coefficient
ω_i	Acentric factor
μ	Viscosity

1. INTRODUCTION

Heavy oil represents approximately half of the world's oil and gas reserves. The development of heavy oil has piqued the interest of many people due to rising global oil consumption. Extra-heavy oil, in particular, has become the focus of attention in the world's oil fields, and its viscosity in formation is typically 10000mPa·s and 50000mPa·s. Cold production technology and hot production technology are the two primary types of heavy oil development technology. Because of the extra-heavy oil's high viscosity, only the thermal production method can effectively be used to develop extra-heavy oil reservoirs [1], [2], [3]. At present, thermal production methods mainly include cyclic steam stimulation (CSS), steam flooding (SF), steam assisted gravity drainage (SAGD), and in-situ combustion (ISC). CCS is the most important approach for developing extra-heavy oil reservoirs among them. The main principle of CCS is to exploit heavy oil by using the characteristics of the viscosity of heavy oil falling rapidly with temperature. In the manufacturing process, it has the advantages of being quick and flexible. However, the production of heavy oil will gradually decrease after numerous stimulation cycles due to the increased water cut and limited sweep area. The oil recovery rate of CCS technology is predicted to be between 10% and 20% [4], [5], [6].

Global climate change is intensifying as CO₂ levels rise, and the natural environment is deteriorating. CO₂ emissions were quite low from the Industrial Revolution

through the middle of the twentieth century. In 1950, the globe emitted 6 billion tons of carbon dioxide, which nearly doubled by 1990 to over 22 billion tons. CO₂ emissions are steadily increasing, with more than 34 billion tons of CO₂ emitted each year. To deal with the crisis and challenges of climate change, CO₂ emissions must be reduced and carbon utilization and storage must be realized [7], [8], [9]. A substantial volume of flue gas is created by steam generators in the process of heavy oil thermal production, which is inconsistent with the concept of sustainable development. Many researchers have discovered that the major components of flue gas produced by fuel burning are CO₂ and N₂, and that injecting these into the formation with steam can considerably improve oil recovery while also lowering carbon emissions, which is an essential CCUS technology. For the first time, flue gas assisted CSS technology was used on a wide scale in China's Bohai Bay Basin, and it was also dubbed multi-thermal-fluid (MTF) huff-puff technology [10], [11].

MTF technology can also improve the efficiency of CSS, allowing extra-heavy oil reservoirs to produce more oil. Solvents, CO₂, N₂, or flue gas co-injection as a thermal supplement are now used to improve the performance of CSS [12], [13], [14]. Flue gas-assisted-CSS is an effective approach to reduce heavy oil viscosity and swelling. Laboratory tests have shown that CO₂-assisted-CSS may reduce heavy oil viscosity by up to 80% compared to CSS [15], [16], [17]. Field applications and laboratory experiments have demonstrated that CO₂-assisted-CSS will increase more oil production than CSS [18], [19], [20], [21].

However, the comprehensive impacts of varied gas mixture ratios on the development of extra-heavy oil reservoirs are not evident due to the multiple components and complex mechanisms of MTF. As a result, it is crucial to investigate the impact of varying CO₂ and N₂ ratios in MTF on extra-heavy oil reserves. In this paper, we firstly used physical experiments to get the components and viscosity of extra-heavy oil sample, and then used the phase equilibrium model and viscosity model to analyze the impact of varying CO₂ and N₂ ratios on the viscosity and phase state of extra-heavy oil. Finally, numerical simulation is used to performed to investigate the impacts of CO₂ and N₂ in MTF, and the ratio of CO₂ and N₂ is optimized to 7:3.

2. EXTRA-HEAVY OIL SAMPLE VISCOSTY TEST

Crude oil data are obtained from the super heavy oil reservoir of M oilfield in Kazakhstan. The reservoir's initial depth, pressure, bubble point pressure, and temperature are 220m, 2.3MPa, 0.38MPa, and 287.4K(14.4°C), respectively. At reservoir conditions, crude oil density, viscosity, volume factor, and

compressibility factor are 0.953g/cm³, 19500mPas, 1.0009, and $1.892 \times 10^{-5} \text{MPa}^{-1}$, respectively. The detailed composition of crude oil is analyzed by gas chromatograph, which is shown in Table 1.

Table 1 Composition of formation crude oil

Components	Molar volume content of surface degassed oil/ mol %	Molar volume content of formation oil/ mol %
CO ₂	0.000	0.013
N ₂	0.000	0.106
C ₁	0.000	2.549
C ₂	0.000	0.003
C ₃	0.000	0.004
iC ₄	0.000	0.002
nC ₄	0.000	0.006
iC ₅	0.000	0.003
nC ₅	0.000	0.005
C ₆	0.000	0.004
C ₇	0.000	0.003
C ₈	0.501	0.489
C ₉	0.182	0.177
C ₁₀	0.288	0.280
C ₁₁	0.986	0.960
C ₁₂	2.150	2.092
C ₁₃	4.064	3.954
C ₁₄	5.276	5.133
C ₁₅	6.800	6.616
C ₁₆	5.845	5.687
C ₁₇	7.147	6.954
C ₁₈	5.536	5.387
C ₁₉₊	61.224	59.571

The viscosity-temperature curve of crude oil samples is obtained by laboratory experiments, which is shown in Fig.1.

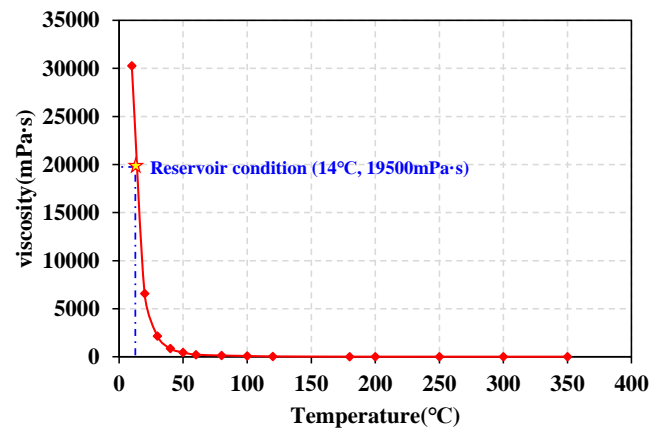


Fig.1 Viscosity-temperature curve of crude oil by physical experiment

3. PHYSICAL PROPERTIES CALCULATION OF EXTRA-HEAVY OIL BEFORE AND AFTER MTF HUFF AND PUFF USING PVT SIMULATION

3.1 Phase equilibrium calculation

Phase equilibrium calculation mainly includes dew-point phase, bubble point phase and isothermal flash phase equilibrium calculation [22]. It is usually calculated by numerical iteration. Because the influence of molecular density on molecular gravity is further considered in gravitational term, the PR(Peng-Robinson) equation is more reasonable in structure, and is also most commonly used in the phase state calculation of hydrocarbon systems in petroleum engineering.

Phase equilibrium calculation should first calculate the initial phase equilibrium constant K_i , Whitson's formula is as follows:

$$K_i = \frac{e^{\left[5.37(1+\omega_i)\left(1-\frac{1}{T_{ri}}\right)\right]}}{P_{ri}} \quad (1)$$

Where, T_{ri} is reduced temperature; P_{ri} is reduced pressure; ω_i is acentric factor.

The PR state equation for multi-component mixed system:

$$p = \frac{RT}{V - b_m} - \frac{a_m(T)}{V(V + b_m) + b_m(V - b_m)} \quad (2)$$

The coefficients a_m and b_m in the formula need to be calculated by van der Waals' mixed rule:

$$a_m(T) = \sum_{i=1}^n \sum_{j=1}^n X_i X_j (a_i a_j \alpha_i \alpha_j)^{0.5} (1 - k_{ij}) \quad (3)$$

$$b_m = \sum_{i=1}^n x_i b_i \quad (4)$$

The deviation coefficient Z cubic equation corresponding to PR equation is as follows:

$$Z_m^3 - (1 - B_m)Z_m^2 + (A_m - 2B_m - 3B_m^2)Z_m - (A_m B_m - B_m^2 - B_m^3) = 0 \quad (5)$$

$$A_m = \frac{a_m(T)p}{(RT)^2} \quad (6)$$

$$B_m = \frac{b_m p}{RT} \quad (7)$$

By substituting the PR equation (2) into the basic thermodynamic equation, the fugacity calculation formula of multi-component mixed system

corresponding to the PR equation can be obtained as follows:

$$\ln\left(\frac{f_i}{x_i p}\right) = \frac{b_i}{b_m}(Z_m - 1) - \ln(Z_m - B_m) - \frac{A_m}{2\sqrt{2}B_m} \left(\frac{2\psi_j}{a_m} - \frac{b_i}{b_m}\right) \ln\left(\frac{Z_m + 2.414B_m}{Z_m - 2.414B_m}\right) \quad (8)$$

$$\psi_j = \sum_j^n x_j (a_i a_j \alpha_i \alpha_j)^{0.5} (1 - k_{ij}) \quad (9)$$

According to the theory of fluid thermodynamic equilibrium equation, when the hydrocarbon-hydrocarbon system reaches gas-liquid equilibrium, the fugacity of each component in the gas and liquid phases are equal, so a new phase equilibrium constant can be calculated:

$$K_i = \frac{y_i}{x_i} = \frac{\frac{f_{il}}{y_i}}{\frac{f_{ig}}{x_i}} \quad (10)$$

Table 2 Thermodynamic properties of each component

Compo nents	Molecula r Weight, kg/kg·m	Specific Gravity	Critical temperature T_{ci} , K	Critical pressure P_{ci} , MPa	Acent ric factor ω_i
CO ₂	44.010	0.818	304.190	7.382	0.228
N ₂	28.013	0.809	126.200	3.460	0.038
C ₁	16.043	0.300	190.560	4.599	0.012
C ₂	30.070	0.356	305.320	4.872	0.100
C ₃	44.097	0.507	369.830	4.248	0.152
iC ₄	58.123	0.563	407.800	3.640	0.184
nC ₄	58.123	0.584	425.120	3.796	0.200
iC ₅	72.150	0.625	460.400	3.380	0.228
nC ₅	72.150	0.631	469.700	3.370	0.252
C ₆	84.000	0.690	512.800	3.330	0.250
C ₇	96.000	0.727	547.200	3.120	0.280
C ₈	107.000	0.749	575.600	2.890	0.312
C ₉	121.000	0.768	602.800	2.640	0.348
C ₁₀	134.000	0.782	626.700	2.420	0.385
C ₁₁	147.000	0.793	647.800	2.240	0.419
C ₁₂	161.000	0.804	668.300	2.080	0.454
C ₁₃	175.000	0.815	686.700	1.970	0.484
C ₁₄	190.000	0.826	705.600	1.860	0.516
C ₁₅	206.000	0.836	724.400	1.830	0.550
C ₁₆	222.000	0.843	740.000	1.660	0.582
C ₁₇	237.000	0.851	755.600	1.590	0.613
C ₁₈	251.000	0.856	766.700	1.530	0.638
C ₁₉₊	475.000	1.100	1060.694	1.550	0.741

3.2 Viscosity calculation

At present, the LBC model and Pedersen model are the most widely used viscosity models in petroleum engineering. the LBC model relates the viscosity at high density with the viscosity at low density and the relative density of the system, and the equation is as follows [23], [24].

$$\left[(\mu - \mu^*) \cdot \zeta + 10^{-4} \right]^{\frac{1}{4}} = \sum_{i=1}^5 a_i \cdot b_r^{i-1} \quad (11)$$

According to Pedersen's model, the viscosity ratio of two fluids with similar properties has a certain functional relationship with their critical temperature ratio, critical pressure ratio, molar mass ratio and coupling coefficient ratio, and the equation is as follows [23], [25]:

$$\mu(p, T) = \left(\frac{T_c}{T_{c0}} \right)^{\frac{1}{6}} \left(\frac{P_c}{P_{c0}} \right)^{\frac{2}{3}} \left(\frac{M_w}{M_{w0}} \right)^{\frac{1}{2}} \frac{\alpha_c}{\alpha_{c0}} \mu_0(p_0, T_0) \quad (12)$$

3.3 Result and discussion

1. Effects of different CO₂ and N₂ ratios on viscosity

As shown in Fig.2, comparing the crude oil viscosity test results with the viscosity calculated by LBC and the viscosity calculated by Pedersen, it can be found that for the viscosity calculation of extra-heavy oil, the LBC method has a large error in calculating the viscosity, and the Pedersen method is more reliable for the viscosity calculation of extra-heavy oil.

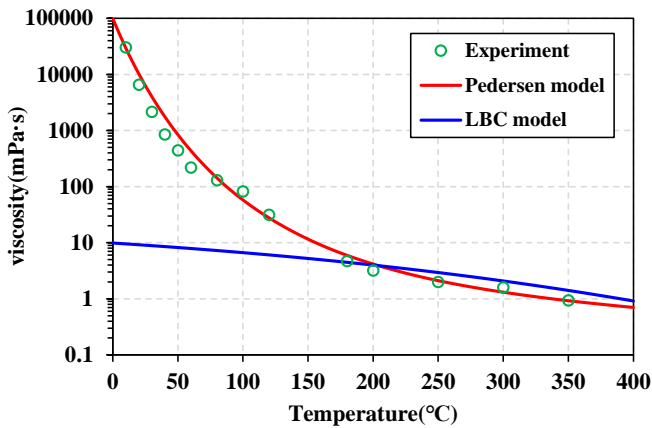


Fig.2 Comparison of the viscosity-temperature curves calculated by the Pedersen and LBC models with those from experimental tests

The ratios of CO₂ and N₂ 1:9, 3:7, 5:5, 7:3 and 9:1 were designed respectively, and the viscosity-temperature curves of different gas mixture ratios were calculated by the Pedersen model, and the fitted gas mixture ratio-viscosity-temperature surfaces are shown

in Fig.3. CO₂ is the main contributor to gas dissolution viscosity reduction, and the higher the content of CO₂, the lower the viscosity of the crude oil after injection. Once the CO₂ ratio exceeds 70%, continued increase of CO₂ has little effect on viscosity reduction of extra-heavy oil.

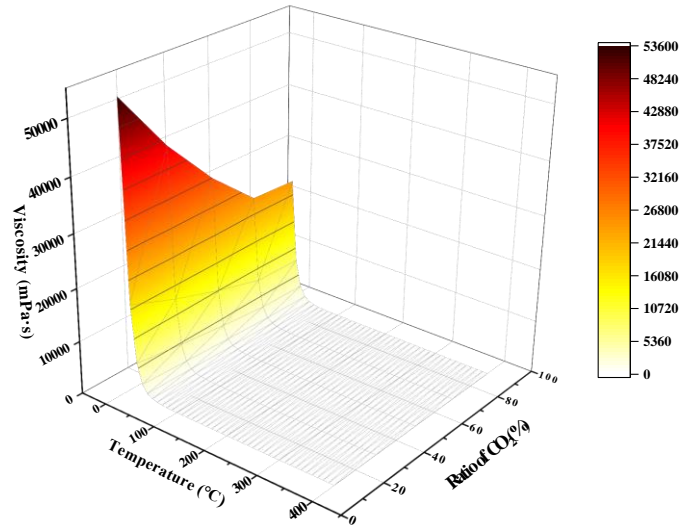


Fig.3 Viscosity-temperature surfaces at different CO₂ ratios

2. Effect of different CO₂ and N₂ ratios on the phase diagram

With the increase of CO₂ ratio and the decrease of N₂ ratio, the phase diagram becomes narrower gradually, and there are obvious changes for the crude oil components and properties after the multi-thermal-fluid throughput.

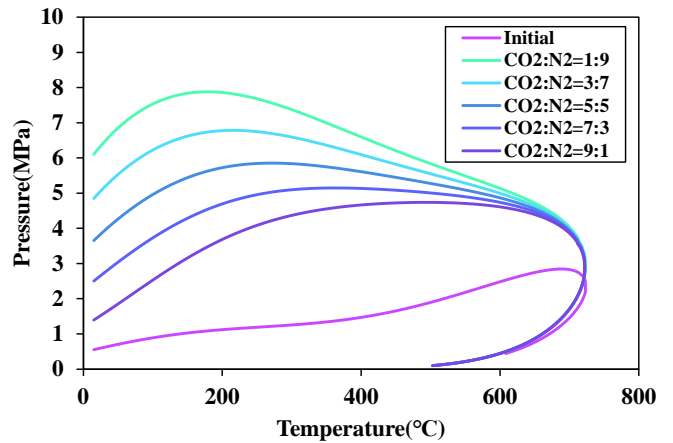


Fig.4 Comparison of the viscosity-temperature curves calculated by the Pedersen and LBC models with those from experimental tests

4. OPTIMIZATION OF RATIO OF CO₂ AND N₂ USING NUMERICAL SIMULATION

Since CO₂ and N₂ in the multi-thermal-fluid(MTF) not only have the effect of viscosity reduction, but also the effect of increasing the formation pressure, where the viscosity reduction is mainly contributed by CO₂, while the effect of replenishing the formation energy is mainly contributed by N₂. Therefore, in order to study the optimal ratio of CO₂ and N₂, a numerical simulation mechanism model was first established using the CMG-STARS module, as shown in Fig. 5. The grid system of the established base model is 50×50×15, with a grid step of 10m×10m×2m; the rock fluid and thermal physical parameters used in the base model are shown in Table 3.

Table 3 Geological and fluid parameters of reservoir

Parameter	Unit	Value
Temperature	°C	16.6
Pressure	MPa	2.6
Rock heat capacity	J/(m ³ ·°C)	2.575×10 ⁶
Rock thermal conductivity	J/(m·d·°C)	1.634×10 ⁵
Water thermal conductivity	J/(m·d·°C)	5.35×10 ⁴
Oil thermal conductivity	J/(m·d·°C)	9.62×10 ³
Rock compressibility	MPa ⁻¹	3.00×10 ⁻⁷
Oil compressibility	MPa ⁻¹	8.28×10 ⁻⁷
Water compressibility	MPa ⁻¹	5.25×10 ⁻⁷
CO ₂ phase equilibrium constant	/	479.17

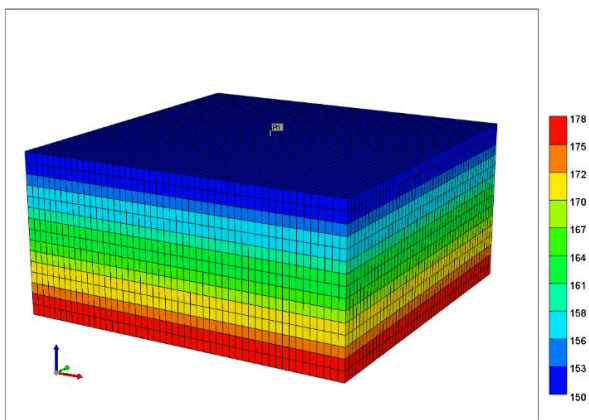


Fig. 5 3D diagram of numerical simulation model

The predicted results from the numerical simulation show that the maximum cumulative oil production is achieved when CO₂ and N₂ are 7:3 (i.e., when the CO₂ ratio is 70%). It indicates that the contribution of

viscosity reduction and energy increase is the largest synergistically.

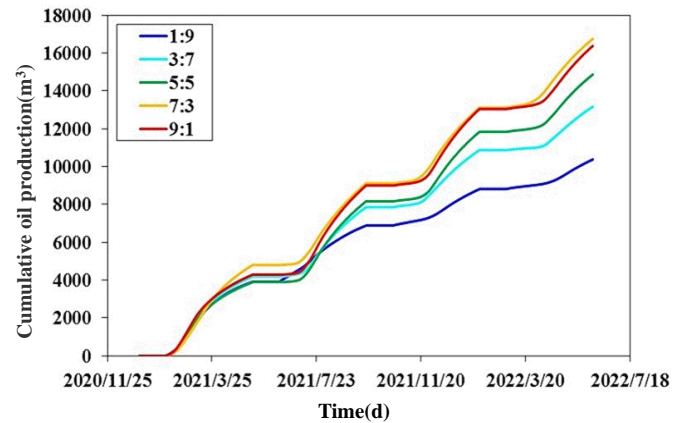


Fig. 6 Cumulative oil production curves under different CO₂ and N₂ ratios

5. CONCLUSION

In order to improve CCUS technology, the use of MTF huff-puff to develop heavy oil is gaining more and more attention. In this paper, the effects of different ratios of CO₂ and N₂ in MTF on extra-heavy oil are researched by PVT simulation and numerical simulation, and the conclusions are as follows:

(1) The Pedersen model is more accurate than the LBC model for calculating heavy oil viscosity, and the viscosity computed by the Pedersen model matches experimental data well.

(2) The higher the ratio of CO₂, the lower the viscosity of the crude oil after MTF huff-puff. With the increase of CO₂ and the decrease of N₂, the phase diagram becomes narrower gradually.

(3) The primary function of CO₂ is to reduce viscosity while the primary function of N₂ is to improve formation energy. The optimal ratio of CO₂ and N₂ is 7:3.

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